



QP CODE: 20000782



20000782

Reg No :

Name :

MSc DEGREE (CSS) EXAMINATION , NOVEMBER 2020

Second Semester

CORE - CH500203 - CHEMICAL BONDING AND COMPUTATIONAL CHEMISTRY

M Sc ANALYTICAL CHEMISTRY, M Sc APPLIED CHEMISTRY , M Sc CHEMISTRY, M Sc
PHARMACEUTICAL CHEMISTRY, M Sc POLYMER CHEMISTRY

2019 Admission Onwards

54E7C630

Time: 3 Hours

Weightage: 30

Part A (Short Answer Questions)

Answer any **eight** questions.

Weight **1** each.

1. How many vibrational modes are present in NH_3 and trans N_2F_2 molecules.
2. How does the departure from cubic symmetry result in a forbidden transition becoming allowed? Explain
3. Write down the Hamiltonian for H_2 molecule
4. What is Hellmann-Feynman theorem.
5. What is Fock operator
6. Write Schrödinger equation for molecules and explain
7. Compare the bond energy, bond length and magnetic behavior of CN and CN^-
8. What is generalized gradient approximation?
9. Write the z-matrix of ammonia molecule.
10. What is CHARMM? Explain its use in molecular mechanics?

($8 \times 1 = 8$ weightage)

Part B (Short Essay/Problems)

Answer any **six** questions.

Weight **2** each.

11. Discuss the IR and Raman activity of trans N_2F_2 molecule
12. What are the orbital selection rules? Explain.
13. What are the important problems faced in quantum mechanical calculations for many particles compared





to a single particle model

14. What are Slater type orbitals (STO) and Gaussian type orbitals (GTO), and sketch STO and GTO
15. Construct the wave functions for CH₄ hybrid orbitals
16. Explain Hückel Molecular Orbital (HMO) theory of allyl system
17. What is meant by SCF procedure? Explain.
18. Distinguish between ab initio and semi empirical methods.

(6×2=12 weightage)

Part C (Essay Type Questions)

Answer any **two** questions.

Weight **5** each.

19. How does group theory help in deducing the hybridisation of BF₃ molecule? Derive the hybrid orbitals for this molecule.
20. Illustrate variation theorem using suitable trial wave function for particle in a one dimensional box
21. Compare and construct MO and VB theories
22. What are the applications of Computational Chemistry?

(2×5=10 weightage)

